Disturbances of Acid Base Equilibrium: Summary

Respiratory Acidosis

Occurs when CO2 is poorly exchanged by the lungs due to inadequate ventilation or perfusion

1. Clinical Findings: ↓ pH ↑ pCO2 ↑ tCO2 (HCO3)

- 2. Causes:
 - a. Lung disease
 - b. Respiratory center injury or depression
 - c. Cardiac disease
- 3. Compensation:
 - a. Kidneys: excrete more H+ and NH4+ and retain HCO3-
 - b. Respiratory system: hyperventilation (if possible)

Respiratory Alkalosis

Occurs when CO2 decreases, usually due to a form of hyperventilation

- 1. Clinical Findings: \uparrow pH \downarrow pCO2 \downarrow tCO2 (HCO3)
- 2. <u>Causes:</u> Hyperventilation
- 3. <u>Compensation:</u>
 - a. Kidneys: retain H+ and NH4+ and excrete more HCO3-
 - b. Respiratory system: hypoventilation (if possible)

Metabolic Acidosis

Occurs when an overproduction of organic acids consumes excess base (i.e. lactic acidosis) or when bicarbonate (base) is lost (diarrhea)

- 1. Clinical Findings: ↓ pH ↓ HCO3-
- 2. <u>Causes</u>:
 - a. Increased production of organic acids and increased anion gap
 - b. Decreased excretion of acids
 - c. Excessive HCO3- loss
- 3. Compensation:
 - a. Respiratory system: hyperventilation
 - b. Kidneys: excrete more H+ and NH4+ and retain more HCO3-

Metabolic Alkalosis

Occurs with a primary HCO3- excess, loss of acid or K+ depletion

- 1. Clinical Findings: ↑pH ↑ HCO3-
- 2. Causes:
 - a. Excessive loss of H+
 - b. Administration of excess alkali
 - c. K+ depletion
- 3. <u>Compensation:</u>
 - a. Respiratory system: hypoventilation
 - b. Kidneys: retain H+ and NH4+ and excrete more HCO3-

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