

Catalase Test

I. PRINCIPLE

A. Reaction Principle

Catalase, an enzyme within the cytochrome enzyme system, is responsible for the decomposition of hydrogen peroxide (H_2O_2) formed during aerobic respiration. All organisms using the cytochrome system of respiration will give a positive catalase reaction when tested. Those organisms using a different system will not produce catalase and will yield a negative reaction.

The mechanism of action is as follows:



B. Clinical Principle

The possession of the catalase enzyme helps to distinguish staphylococci from streptococci and is useful in the identification of other bacteria.

II. REAGENTS

A. Reagents and Supplies

1. 3% Hydrogen peroxide solution
 - a. Store H_2O_2 at $4 \pm 2^\circ C$.
 - b. Do not expose H_2O_2 to light, heat or dust.
 - c. Mark the date opened on the bottle and record a 6 month outdate, from the date opened. Do not exceed manufacturer's outdate.
2. Glass slide
3. Wooden applicator stick or Bio-loop (do not use nickel loops)

III. INTERFERING SUBSTANCES

1. When picking up a bacterial colony, be careful not to pick up any blood agar. The red cells contain catalase and will give a false positive reaction.
2. When performing a catalase test on an isolate from a blood culture, do not select the colony from the first quadrant. The presence of blood will produce a false result.
3. Do not use nickel loops that will give a false positive reaction.
4. Do not use H_2O_2 that is past the expiration date.

IV. QUALITY CONTROL

- A. Quality control is performed each new batch, lot number, and shipment.
- B. Staph sp. is used as the positive control and a Strep sp. is used as the negative control.
- C. Expected results:
 1. Staph spp.: bubbling reaction

- 2. Strep sp.: absence of bubbling reaction
- D. Controls must display expected results. If controls do not display expected results, quality control must be repeated.
- E. Record results in the computer.

V. PROCEDURE

- A. Place one drop of H₂O₂ on a glass slide.
- B. Pick up the bacterial colony to be tested with a wooden applicator stick or a Bio-loop.
- C. Suspend the colony in the drop of H₂O₂.
- D. Observe the reaction over a dark background.

VI. INTERPRETATION

- A. A positive test is a bubbling reaction caused by the release of O₂ from the H₂O₂ in the presence of catalase.
- B. A negative test is the absence of any bubbling reaction.

VII. NOTES

- A. Do not use expired reagent.
- B. Care should be taken when working with a H₂O₂ solution. Do not splash. It may cause irritation to the skin or eyes. Rinse affected areas with water upon exposure.

VIII. REFERENCES

- A. Burrows, L., "Textbook of Microbiology". Philadelphia, W.B. Saunders Company, 127-128, 1963.
- B. Blazevic and Ederer, Principles of Biochemical Tests in Diagnostic Microbiology, 13-14, 1975.

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